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# Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

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|   | Application No.  | Applicant(s)   |
|---|--|--|
|   | 10/528,813   | ZANIBELLI ET AL.   |
| Office Action Summary   | Examiner   | Art Unit   |
|   | PREM C. SINGH  | 1797   |
| The MAILING DATE of this communication ap<br>Period for Reply   | pears on the cover sheet with the c  | orrespondence address  |
| A SHORTENED STATUTORY PERIOD FOR REPL WHICHEVER IS LONGER, FROM THE MAILING D  - Extensions of time may be available under the provisions of 37 CFR 1. after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory period  - Failure to reply within the set or extended period for reply will, by statut-Any reply received by the Office later than three months after the mailin earned patent term adjustment. See 37 CFR 1.704(b). | DATE OF THIS COMMUNICATION 136(a). In no event, however, may a reply be tin will apply and will expire SIX (6) MONTHS from e, cause the application to become ABANDONE | N. nely filed the mailing date of this communication. D (35 U.S.C. § 133). |
| Status  |  |  |
| Responsive to communication(s) filed on <u>08 F</u> This action is <b>FINAL</b> . 2b) ☑ This      Since this application is in condition for allowated closed in accordance with the practice under the practice under the practice.  | s action is non-final.<br>ance except for formal matters, pro  |  |
| Disposition of Claims   |  |  |
| 4)  Claim(s) 1-49 is/are pending in the application 4a) Of the above claim(s) 31-49 is/are withdra 5)  Claim(s) is/are allowed. 6)  Claim(s) 1-30 is/are rejected. 7)  Claim(s) is/are objected to. 8)  Claim(s) are subject to restriction and/o Application Papers 9)  The specification is objected to by the Examine 10)  The drawing(s) filed on 22 March 2005 is/are: Applicant may not request that any objection to the   | wn from consideration. or election requirement. er. a)⊠ accepted or b)⊡ objected to  | · ·  |
| Replacement drawing sheet(s) including the correct 11) The oath or declaration is objected to by the E  | ction is required if the drawing(s) is ob  | jected to. See 37 CFR 1.121(d).  |
| Priority under 35 U.S.C. § 119  |  |  |
| 12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of:  1. Certified copies of the priority documen 2. Certified copies of the priority documen 3. Copies of the certified copies of the priority application from the International Burea * See the attached detailed Office action for a list   | ts have been received.<br>ts have been received in Applicationity documents have been receive<br>nu (PCT Rule 17.2(a)).  | on No ed in this National Stage  |
| Attachment(s)  1) Notice of References Cited (PTO-892)  2) Notice of Draftsperson's Patent Drawing Review (PTO-948)  3) Information Disclosure Statement(s) (PTO/SB/08)  Paper No(s)/Mail Date 01/26/07; 05/16/05; 03/23/05.  | 4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal F 6) Other:   | ate  |

### **DETAILED ACTION**

#### Election/Restrictions

1. Group I, claim(s) 1-30, drawn to a process for oxidative desulphurization.

Group II, claim(s) 31-42, drawn to a catalyst composition.

Group III, claim(s) 43-49, drawn to a method for making a catalyst composition.

The inventions listed as Groups I-III do not relate to a single general inventive concept under PCT Rule 13.1 because, under PCT Rule 13.2, they lack the same or corresponding special technical features for the following reasons: The corresponding technical feature shared by all of the groups is the catalyst composition.

The inventions are distinct, each from the other because of the following reasons:

Inventions in Group III and II are related as process of making and product made.

The inventions are distinct if either or both of the following can be shown: (1) that the process as claimed can be used to make another and materially different product or (2) that the product as claimed can be made by another and materially different process (MPEP § 806.05(f)). In the instant case the process can be used to make another and materially different product, for example a hydrodesulfurization catalyst.

Inventions II and I are related as product and process of use. The inventions can be shown to be distinct if either or both of the following can be shown: (1) the process for using the product as claimed can be practiced with another materially different

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product or (2) the product as claimed can be used in a materially different process of using that product. See MPEP § 806.05(h). In the instant case the process can be conducted by a materially different product, for example, a hydrodesulfurization catalyst.

Inventions in Group I and III are unrelated. Inventions are unrelated if it can be shown that they are not disclosed as capable of use together and they have different designs, modes of operation, and effects (MPEP § 802.01 and § 806.06). In the instant case, the process of oxydesulfurization and the process of making a catalyst composition are unrelated because they have different designs, mode of operation, feed, operating conditions, and end products.

International Preliminary Examination report submitted by the Applicant indicated that the catalyst compound is an obvious variant over prior art cited in the international application.

Accordingly, because the corresponding technical feature is obvious or not novel, and the claims in Group I, II and III are drawn to different inventions, the restriction is proper.

## Claim Rejections - 35 USC § 112

2. The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

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3. Claim 27 is rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

Regarding claim 27, the phrase "unique" renders the claim indefinite because it is unclear whether the limitations following the phrase are part of the claimed invention.

See MPEP § 2173.05(d).

# Claim Rejections - 35 USC § 103

4. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

The factual inquiries set forth in *Graham* **v.** *John Deere Co.*, 383 U.S. 1, 148 USPQ 459 (1966), that are applied for establishing a background for determining obviousness under 35 U.S.C. 103(a) are summarized as follows:

- 1. Determining the scope and contents of the prior art.
- 2. Ascertaining the differences between the prior art and the claims at issue.
- 3. Resolving the level of ordinary skill in the pertinent art.
- 4. Considering objective evidence present in the application indicating obviousness or nonobviousness.

This application currently names joint inventors. In considering patentability of the claims under 35 U.S.C. 103(a), the examiner presumes that the subject matter of the various claims was commonly owned at the time any inventions covered therein

were made absent any evidence to the contrary. Applicant is advised of the obligation under 37 CFR 1.56 to point out the inventor and invention dates of each claim that was not commonly owned at the time a later invention was made in order for the examiner to consider the applicability of 35 U.S.C. 103(c) and potential 35 U.S.C. 102(e), (f) or (g) prior art under 35 U.S.C. 103(a).

- 5. Claims 1-30 are rejected under 35 U.S.C. 103(a) as being unpatentable over Cabrera et al (US Patent 6,171,478) in view of Rabion et al, WO 01/48119 A1).
- 6. With respect to claim 1, Cabrera discloses a process for oxidative desulfurization of hydrotreated hydrocarbon mixtures boiling in a range of 149 to 538°C, containing 100-1000 wppm sulfur (See column 3, lines 20-38; column 4, lines 38-43; column 5, lines 9-11), which comprises:
- (a) putting these mixtures in contact, in the presence of any suitable known aqueous oxidizing solution like acetic acid and hydrogen peroxide (See column 5, lines 55-62; column 6, lines 9-10), with a catalytic composition comprising a supported transition metal, a transition metal complex or a transition metal oxide (See column 6, lines 13-18); and
- (b) separating the obtained corresponding sulfur oxygenated products from the hydrocarbon mixture (See column 3, lines 32-38).

It is to be noted that Cabrera invention uses the catalyst composition in a separate step after oxidative desulfurization step.

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Rabion discloses a process similar to Cabrera for oxidative desulfurization of hydrocarbon mixtures, including gasoline, kerosene and diesel containing thiophenic sulfur (See page 2, paragraph 2, page 4, paragraph 1). Rabion also discloses using organic peroxide, hydrogen peroxide and peracids along with a catalytic composition formed by titanium, zirconium, vanadium, chromium, molybdenum, tungsten, iron, and manganese supported on silica, alumina, zirconia, aluminosilicates crystalline or amorphous and the aluminophosphates mesoporous solids, alone or in combination (See page 3, paragraph 2). Rabion further discloses the synergistic effect of combined action of oxidizing agents and the catalysts used (See page 3, line 1).

Thus, it would have been obvious to one skilled in the art at the time of invention to invention to modify Cabrera process and use the catalyst mixed with the oxidizing solution to simplify the recovery and separation of the reaction products including sulfur-oxidated compounds recovered from the oxidation zone and synergistic effect of combined oxidizing agents and catalysts used (See Cabreara, column 6, lines 18-21; Rabion, page 3, line 1).

Cabrera invention does not specifically disclose the claimed oxide matrix, however, the invention does disclose using supported transition metals. Rabion discloses the claimed transition metals as well as oxide matrix (See Rabion, page 3, paragraph 2).

7. With respect to claim 2, Cabrera invention does not specifically disclose that the surface of the catalytic composition has  $-O - Si(R)_3$  groups. However, Rabion

discloses catalyst support similar to the Applicant's claims (See Rabion, page 3, paragraph 2). Thus, it is expected that the catalytic composition of Rabion invention should necessarily have the claimed groups.

8. With respect to claims 3, 4, 8 and 9, Cabrera invention discloses using transition metal oxides which includes Ti, V, Zr and their mixtures (See column 6, lines 16-19)..

Although Cabrera invention does not specifically disclose using boron and/or gallium, it is known to those skilled in the art that boron and gallium are widely used as sulfur reduction additives (evidenced by Wormbecher et al., US Patent 5,376,608: column 1, lines 65-68; column 2, lines 1-2). Thus, it would have been obvious to one skilled in the art at the time of invention to modify Cabrera invention and add boron and/or gallium in the catalyst to enhance sulfur reduction in the process.

- 9. With respect to claims 5-7 and 15, modified Cabrera invention does not specifically disclose molar ratio of oxide matrix and the metal oxide in the catalytic composition, however, since the molar ratio of oxide matrix and the metal oxide is a result-effective variable, it would have been obvious to one skilled in the art at the time of invention to modify Cabrera invention and determine the optimum ratio for an effective desulfurization process. See *In re Boesch*, 617 F.2d 272, 205 USPQ 215 (CCPA 1980).
- 10. With respect to claims 10-14, modified Cabrera invention does not specifically disclose the physical characteristics of the catalytic composition. Since Rabion invention

is using a catalyst similar to the Applicant's claim, it is expected that the physical properties of Rabion catalyst will inherently have similar physical characteristics as the claimed catalyst.

- 11. With respect to claim 16, Rabion invention does not specifically disclose catalytic composition as an extrudate, however, the invention does disclose using catalyst in the solid form (See Table 2) which should necessarily have been prepared by standard known techniques, including extrusion. Thus, it would have been obvious to one skilled in the art at the time of invention to use catalyst extrudates because any form and shape of the catalyst is expected to be equally effective.
- 12. With respect to claims 17 and 18, Cabrera invention discloses that any suitable known aqueous oxidizing solution, including hydrogen peroxide, may be used to perform the sulfur oxidation (See column 5, lines 59-62). Rabion invention discloses that any organic peroxide, hydrogen peroxide and organic hydroperoxides may be used (See page 3, paragraph 1). Thus, it would have been obvious to one skilled in the art at the time of invention to use any organic peroxide, including the claimed peroxides because they are all expected to be equally effective in the oxidative desulfurization process.
- 13. With respect to claim 19, Cabrera invention discloses that the process is carried out at a temperature between 38 and 149°C (See column 6, lines 3-5).

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14. With respect to claims 20-22, Rabion invention does not specifically disclose the molar ratio of organic peroxide and sulfur, however the invention does disclose using 0.5 mmol of benzothiophene, 0.5 mmol of dibenzothiophene, 20 ml of n-decane and 1 ml of aqueous solution of 30 wt% hydrogen peroxide in an experiment (See page 5, paragraph 3). Thus, it would have been obvious to one skilled in the art at the time of invention to modify Rabion invention and determine the molar ratio of organic peroxide and sulfur. It is expected that the ratio will be in a rage similar to the claimed range because Rabion is using a similar catalyst to oxidize thiophenes at the operating conditions similar to the claimed conditions.

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- 15. Claims 23 and 24 have all the limitations of claims 19-22, and discussed above.
- 16. With respect to claim 25, Cabrera invention discloses that the process is carried out at atmospheric pressure (See column 6, lines 3-4).
- 17. With respect to claim 26, Cabrera invention discloses that the hydrotreated hydrocarbon mixture contains hydrocarbons boiling in the range of 93 to 565°C (See column 4, lines 38-43). Thus, it would have been obvious to one skilled in the art at the time of invention to use light cycle oil (LCO) as claimed, because its boiling range falls under Cabrera disclosure.

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18. With respect to claim 27, Cabrera invention discloses that the process is carried

out in an organic phase and in the absence of added solvent (See column 3, lines 20-

38).

19. With respect to claims 28 and 29, Cabrera invention discloses that the separation

of the obtained sulfur oxygenated products from the hydrotreated hydrocarbon mixtures

is carried out by solvent extraction using any suitable known selective solvent including

acetonitrile (See column 6, lines 35-39).

20. With respect to claim 30, Rabion invention does not specifically disclose the

molar ratio of silica to alumina, however, the invention does disclose using silica and

alumina in combination (See page 3, paragraph 2). Since the molar ratio of silica to

alumina in the mixed oxide matrix is a result-effective variable, it would have been

obvious to one skilled in the art at the time of invention to modify Rabion invention and

determine the optimum ratio for an effective desulfurization process. See *In re Boesch*,

617 F.2d 272, 205 USPQ 215 (CCPA 1980).

Conclusion

21. The prior art made of record and not relied upon is considered pertinent to

applicant's disclosure.

Herbstman et al (US Patent 3,565,793).

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Any inquiry concerning this communication or earlier communications from the examiner should be directed to PREM C. SINGH whose telephone number is (571)272-6381. The examiner can normally be reached on 7:00 AM to 3:30 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Glenn Caldarola can be reached on 571-272-1444. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

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